

signals for OH or NH groups, but showed a carbonyl band at  $1700\text{ cm}^{-1}$ . The mass spectrum exhibited the parent peak at  $m/e$  340 and primary fragmentations parallel to and the corresponding ion peaks 30 mu higher than those of 3a. An M-31 peak at  $m/e$  309 indicated the presence of a carbomethoxy group and the peak at  $m/e$  282 could be rationalised by assuming successive loss of a CO molecule and two methyl radicals from the parent ion. The mass spectrum thus suggested that D contains an additional methoxy group compared to 3a. The above data coupled with the co-occurrence with aristolochic acid-D (1b), pointed to structure 3f for compound D. The structure was finally confirmed as follows: denitration of aristolochic acid-D (1b) with  $\text{NaBH}_4$  (cf. 1a  $\rightarrow$  3b [2]) yielded 3g which on treatment with  $\text{CH}_2\text{N}_2$  and chromatographic purification furnished the major product identical (TLC, IR, mmp) with the natural compound. On saponification, 3f furnished the hitherto unreported aristolinic acid (3h), mp  $284\text{--}285^\circ$  ( $\text{CHCl}_3\text{--MeOH}$ ).

Compound E, encountered in petrol and benzene extracts, was crystallised from  $\text{CHCl}_3\text{--MeOH}$  as yellow needles, mp  $285\text{--}286^\circ$ . IR ( $\nu_{\text{max}}^{\text{Nujol}}$   $1700, 1510, 1340\text{ cm}^{-1}$ ) indicated it to be an aromatic nitro compound contain-

ing a carbonyl function. The mass spectrum suggested the molecular formula  $\text{C}_{18}\text{H}_{13}\text{NO}_7$  ( $M^+$  355) and a peak at  $M - 46$  supported the presence of a nitro group. The compound was identified as methyl aristolochate (1e) by direct comparison (TLC, IR, mmp) with an authentic specimen.

**Plant material.** *Aristolochia indica* L. (Aristolochiaceae). A voucher specimen identified by the National Botanical Gardens (Calcutta) is available in the herbarium of the suppliers, Messrs United Chemicals and Allied Products, Calcutta, Kerala (South India).

**Acknowledgements.**—The authors are indebted to Prof. H. Furukawa of Meijo University, Nagoya, Japan for confirming the identity of methyl aristolate, and to Dr. U. R. Ghatak of the Indian Association for the Cultivation of Science, Calcutta for the NMR spectra. Thanks are also due to our colleague Dr. E. Ali for the mass spectra.

#### REFERENCES

1. Kupchan, S. M. and Merianos, J. J. (1968) *J. Org. Chem.* **10**, 3735.
2. Ito, K., Furukawa, H. and Haruna, M. (1972) *Yakugaku Zasshi* **92**, 92.

*Phytochemistry*, 1977, Vol. 16, pp. 1104–1105. Pergamon Press. Printed in England.

### NEW DIBENZOYLMETHANE AND CHALCONE DERIVATIVES FROM *MILLETIA OVALIFOLIA* SEEDS

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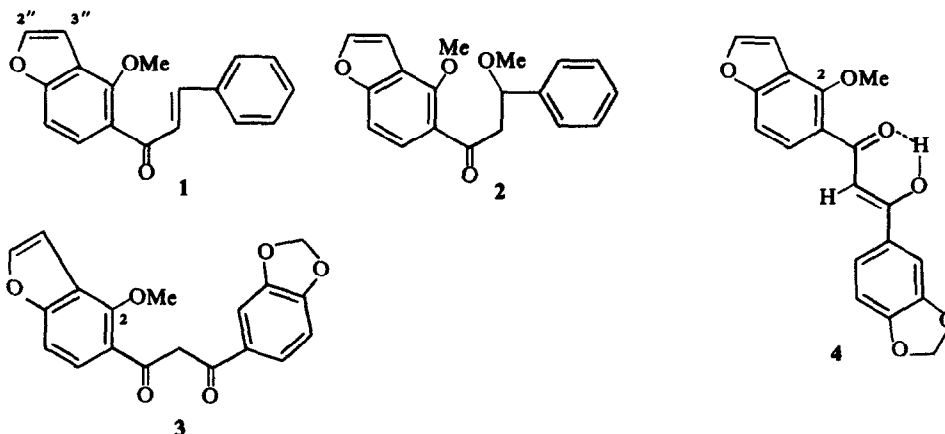
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(Received 28 January 1977)

**Key Word Index.**—*Milletia ovalifolia*; Leguminosae; ovalitenin A; ovalitenin B; ovalitenone.

In continuation of our earlier work [1], further examination of *Milletia ovalifolia* has led to the isolation of three more new compounds. Ovalitenin A:  $\text{C}_{18}\text{H}_{14}\text{O}_3$ ;  $M^+$  (278), mp  $99\text{--}100^\circ$ , pale yellow needles. +ve Mg-HCl test. UV (MeOH)  $\lambda_{\text{max}}$  log  $\epsilon$  245 (4.17), 310 (4.33). IR (KBr)  $\nu$ : 1640, 1590, 1500, 1350, 1250, 1200, 1150, 1080, 1050, 980, 815,  $760\text{ cm}^{-1}$ . PMR ( $\delta$  value, solvent  $\text{CDCl}_3$ ): Methoxyl and furan protons shown by peaks at 4.15 (s, 3H 2'-OMe), 7.13 (d, 1H,  $J = 2\text{ Hz H-3''}$ ), 7.73 (d,  $J = 2\text{ Hz H-2''}$ ), aromatic protons and  $\alpha, \beta$  protons appeared as follows; 7.54 (d, 1H,  $J = 10\text{ Hz, H-6'}$ ) and 7.28–7.54 (m, 8H, H-5', 2, 3, 4, 5, 6,  $\alpha$  and  $\beta$ ). MS showed fragments at  $m/e$  263 (65.5), 175 (99.5), 161 (98), 160 (99), 131 (83), 104 (17), 103 (98), 77 (83). These data and the cooccurrence with pongamol in the same plant led to structure 1 for ovalitenin A. This was confirmed by synthesis from 2-hydroxy-(4',5'-3,4)-furanoacetophenone which was prepared from 3-allylresacetophenone by  $\text{OsO}_4\text{--KIO}_4$  oxida-

tion followed by cyclization with *o*-phosphoric acid mp  $85^\circ$  (lit. [2] mp  $86^\circ$ ), UV (MeOH)  $\lambda_{\text{max}}$  nm (log  $\epsilon$ ): 235 (4.41), 275 (3.67), 325 (3.37). IR (KBr)  $\nu$ :  $1640\text{ cm}^{-1}$ . PMR ( $\delta$ ,  $\text{CDCl}_3$ ): 2.59 (s, 3H, —COMe), 6.97 (d, 1H,  $J = 2\text{ Hz, H-3'}$ ), 7.05 (d, 1H,  $J = 9\text{ Hz, H-5}$ ), 7.55 (d, 1H,  $J = 2\text{ Hz H-2'}$ ), 7.65 (d, 1H,  $J = 9\text{ Hz, H-6}$ ), 13.79 (s, 1H, 2-OH). Methylation with  $(\text{Me})_2\text{SO}_4\text{--K}_2\text{CO}_3\text{--Me}_2\text{CO}$  gave 2-methoxy-(4',5'-3,4)-furanoacetophenone mp  $59\text{--}60^\circ$  (lit. [2] mp  $59^\circ$ ). IR (KBr)  $\nu$ :  $1635\text{ cm}^{-1}$ . PMR ( $\delta$ ,  $\text{CDCl}_3$ ): 2.69 (s, 3H, —COMe), 4.1 (s, 3H, 2-OMe), 6.93 (d, 1H,  $J = 2\text{ Hz, H-3'}$ ), 7.05 (d, 1H,  $J = 9\text{ Hz, H-5}$ ), 7.55 (d, 1H,  $J = 2\text{ Hz, H-2'}$ ), 7.76 (d, 1H,  $J = 9\text{ Hz, H-6}$ ). The above ketone on condensation with benzaldehyde under alkaline conditions yielded a chalcone which was identical with the natural sample (coTLC, mmp, coIR in KBr). Ovalitenin B:  $\text{C}_{19}\text{H}_{18}\text{O}_4$ ;  $M^+$  (310), mp  $77\text{--}78^\circ$ , colourless needles. +ve Mg-HCl test. UV (MeOH)  $\lambda_{\text{max}}$  nm log  $\epsilon$ : 235 (4.62), 275 (3.66), 305 (3.36); IR (KBr)  $\nu$ :



1660, 1595, 1580, 1465, 1420, 1360, 1250, 1205, 1105, 1070, 1000, 960, 850, 750, 700  $\text{cm}^{-1}$ . PMR ( $\delta$  value  $\text{CDCl}_3$ ) showed one aliphatic methoxyl at 3.21 (s, 3H,  $\beta$ -OMe) and aromatic methoxyl at 4.1 (s, 3H, 2'-OMe), other peaks at 3.4 (m, 2H,  $-\text{CO}-\text{CH}_2-$ ), 4.85 (m, 1H,  $\text{MeO}-\text{CH}-\text{Ph}$ ). Furan and aromatic protons appeared at 6.98 (d, 1H,  $J = 2$  Hz, H-3'), 7.07 (d, 1H,  $J = 9$  Hz, H-5'), 7.4 (s, 5H, H-2, 3, 4, 5, 6), 7.6 (d, 1H,  $J = 2$  Hz, H-2') and 7.7 (d, 1H,  $J = 9$  Hz, H-6'). MS showed fragments at  $m/e$  295 (23), 278 (7), 263 (27), 175 (69), 174 (99), 161 (22.5), 161 (88), 148 (33), 132 (21), 131 (22), 121 (100), 105 (73), 77 (94). The cooccurrence with pongamol and the mass fragment ion peak at 121 (100) required aliphatic  $-\text{OMe}$  at  $\beta$ -position and led to structure 2 for ovalitenin B. This was confirmed by converting it into ovalitenin A by elimination of MeOH using conc  $\text{H}_2\text{SO}_4$  at room temperature. Ovalitenone  $\text{C}_{19}\text{H}_{14}\text{O}_6$   $\text{M}^+$  (338), mp 119–20°, yellow crystals, +ve Mg-HCl and Labat tests. UV (MeOH)  $\lambda_{\text{max}}$  nm (log  $\epsilon$ ): 240 (4.46), 260 (4.22), 365 (4.41). IR (KBr)  $\nu$ : 1600, 1580, 1520, 1460, 1350, 1260, 1030, 930, 790, 730  $\text{cm}^{-1}$ . PMR ( $\delta$ , values,  $\text{CDCl}_3$ ) showed one methoxyl and methylenedioxy groups at 4.02 (s, 3H, 2-OMe), 6.0 (s, 2H, 3',4'-O- $\text{CH}_2$ -O). Furan protons appeared at 6.83 (d, 1H,  $J = 2$  Hz-3') and 7.45 (d, 1H,  $J = 2$  Hz, H-2'), aromatic and olefinic proton due to enolic form (4) appeared as

follows: 6.73 (d, 1H,  $J = 9$  Hz, H-5), 6.86 (d, 1H,  $J = 8$  Hz, H-5'), 7.05–7.37 (m, 3H, H-2', 6' and one olefinic proton), 7.7 (d, 1H, 1H,  $J = 9$  Hz, H-6). Two sharp singlets due to labile protons at 3.91 (s,  $\frac{1}{2}\text{H}$ ) and 4.47 (s,  $\frac{1}{2}\text{H}$ ), whose intensities decreased after  $\text{D}_2\text{O}$  exchange, assigned to the methylene proton of dibenzoylmethane and olefinic proton of enolic form [3]. The MS showed fragments at  $m/e$  307 (100), 320 (33.5) due to (M-31) and (M-18) and others at 291 (4), 175 (99), 160 (52.5), 149 (99), 148 (25), 133 (13.5), 132 (16), 122 (23). These data led to structure 3 for ovalitenone. This was confirmed by absolute MeOH-KOH cleavage to 2-methoxy-(4', 5',-3,4)-furanacetophenone and piperonylic acid and by HI-Ac<sub>2</sub>O demethylation followed by cyclization to pongaglabrone, identical with the authentic samples.

**Acknowledgements**—The authors are grateful to the Director, CIBA-Geigy Research Centre, Bombay for the MS and to the Council of Scientific and Industrial Research (India) for financial assistance to one of them (R.K.G.).

#### REFERENCES

1. Gupta, R. K. and Krishnamurti, M. (1976) *Phytochemistry* **15**, 832, 1795.
2. Narayanaswami, S., Rangaswami, S. and Seshadri, T. R. (1954) *J. Chem. Soc.* 1871.
3. Khan, H. and Zaman, A. (1974) *Tetrahedron* **30**, 2811.